Highly Enantioselective Asymmetric Autocatalysis of Pyrimidin-5-yl Alkanol Induced by Chiral 1,3-Disubstituted Hydrocarbon Allenes

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Dedicated to Professor Dieter Seebach on the occasion of his 65th birthday

1,3-Disubstituted chiral allenes without any heteroatoms act as chiral initiators in the addition of $(i-Pr)_2Zn$ to pyrimidine-5-carbaldehyde to afford, in combination with the subsequent asymmetric autocatalysis, chiral pyrimidin-5-yl alkanols with up to 98% ee. The absolute configuration of the pyrimidin-5-yl alkanol formed depend on that of the chiral allene.

Introduction. – 1,3-Disubstituted allene is one of the representative chiral compounds without any stereogenic center [1]. Two consecutive π faces are twisted vertically to locate the terminal substituent on the *re* or *si* face of another π face. Recent progress in the synthesis of allenes made it possible to obtain optically active allenes by enantioselective synthesis [2–4]. Hence, applications of optically active allenes as synthetic intermediates have been reported [5]. However, to the best of our knowledge, the use of chiral allenes as chiral catalysts or chiral ligands has rarely been reported.



 $R^1, R^2 \neq H$

Meanwhile, during our continuing study of asymmetric autocatalysis [6], it was found that asymmetric autocatalysis of pyrimidin-5-yl alkanol in the addition of (i-Pr)₂Zn to pyrimidine-5-carbaldehyde proceeds with amplification of ee [6] (for reviews on asymmetric autocatalysis with amplification of ee, see [7a-f]; for reviews including both autocatalytic and non-autocatalytic reactions with amplification of ee, see [7gi]). Moreover, when (i-Pr)₂Zn was reacted with pyrimidine-5-carbaldehyde in the presence of chiral initiators such as amino acids, helicenes, deuterated primary alcohols,

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quartz, and sodium chlorate, the absolute configuration of the pyrimidin-5-yl alkanol obtained depends on that of the chiral initiator [8].

We report herein the asymmetric autocatalysis of pyrimidin-5-yl alkanol in the presence of 1,3-disubstituted chiral allenes. Both enantiomers of highly enantiomerically enriched pyrimidin-5-yl alkanols were obtained from axially chiral allenes as the sole chiral source.

Results and Discussion. – Enantioselective addition of (i-Pr)₂Zn to 2-(alkynyl)pyrimidine-5-carbaldehyde 1 in the presence of chiral allenes $2\mathbf{a} - \mathbf{e}$ were examined (Scheme). To an ice-cooled methylcyclohexane solution of 1 and chiral allene 2, a hexane solution of (i-Pr)₂Zn was slowly added. The solution was then diluted with toluene, and aldehyde 1 and $(i-Pr)_2Zn$ were added portionwise. Aqueous workup gave enantiomerically enriched 2-(alkynyl)pyrimidin-5-yl alkanol 3. The results are summarized in the *Table*. As shown in the *Scheme*, when (+)-(S)-1,3-diphenylpropadiene **2a** was used as a chiral initiator, (R)-pyrimidin-5-yl alkanol **3** with 97% ee was obtained in 95% isolated yield (Entry 1). On the other hand, when the other enantiomer of chiral allene (-)-(R)-**2a** was used as chiral initiator, (S)-pyrimidin-5-yl alkanol **3** with 98% ee was obtained in 95% isolated yield (*Entry 2*). Thus, the absolute configuration of the pyrimidin-5-yl alkanol **3** obtained was dependent on that of the chiral allene. Toluene as a solvent (*Entries 3* and 4) gave results similar to those with methylcyclohexane as solvent (*Entries 1* and 2). The reaction in the presence of (+)- or (-)-1cyclohexyl-3-phenylpropadiene (2b) gave (R)- and (S)-pyrimidin-5-yl alkanol 3 with 97% ee, respectively (Entries 5 and 6). According to the Lowe's experimental rule and the subsequent studies based on circular dichroism, the absolute configuration of 1phenylallenes and 1,3-dialkylallenes have been correlated with the sign of specific rotation [3a][9].

Scheme. Enantioselective Addition of $(i-Pr)_2$ / Zn to 2-(Alkynyl)peprinidine-5-carbaldehyde **1** in the Presence of Chiral Allenes **2a**-e



Therefore, allenes with *dextro* rotation may have the (S) configuration similar to the 1,3-diphenylpropadiene [10]. The derivatives of 1-phenylallenes were also used as chiral initiators in asymmetric autocatalysis. When chiral (+)-(S)-4-methyl-1-phenylpenta-1,2-diene **2c** [3a] was used as a chiral initiator, (R)-pyrimidin-5-yl alkanol **3** with 96% ee was formed in a yield of 96% (*Entry* 7). On the other hand, (S)-pyrimidin-5-yl alkanol with 94% ee was obtained in a yield of 94% in the corresponding reaction with (-)-(R)-**2c** as a chiral initiator (*Entry* 8). Similarly, chiral (+)- and (-)-1-phenylallene derivatives **2d** with PhCH₂ substituents gave (R)- and (S)-pyrimidin-5-yl alkanol **3** with 94 and 95% ee, respectively (*Entries* 9 and 10). In addition, nonconjugated allene was used as a chiral initiator. (+)-1,5-Diphenylpentadiene (**2e**), *i.e.*, a chiral allene with two PhCH₂ substituents, afforded (R)-**3** (*Entry* 11), whereas the other enantiomer (-)-**2e** gave (S)-**3** (*Entry* 12).

Entry	Allene 2					Pyrimidin-5-yl alkanol 3	
		\mathbb{R}^1	\mathbb{R}^2	ee/% ^b)	$[\alpha]_{\rm D}$ (config.)	Yield/%	ee/% (config.)
1°)	2a	Ph	Ph	> 99.5	+ (S)	95	97 (<i>R</i>)
$2^{\rm c}$)				> 99.5	-(R)	95	98(S)
3 ^d)				> 99.5	+(S)	92	93 (<i>R</i>)
$4^{\rm d}$)				> 99.5	-(R)	94	94(S)
5	2b	Ph	Cyclohexyl	90	+	88	97 (<i>R</i>)
6				94	_	90	97(S)
7	2c	Ph	i-Pr	91	+ (S)	96	96 (<i>R</i>)
8				> 99.5	-(R)	94	94 (S)
9	2d	Ph	$PhCH_2$	43	+	96	94 (<i>R</i>)
10			-	56	_	92	95 (S)
11	2e	CH_2Ph	$PhCH_2$	97	+	97	90 (<i>R</i>)
12		_	-	92	_	97	97 (<i>S</i>)

 $Table. \ Highly \ Enantioselective \ Synthesis \ of \ Pyrimidin-5-yl \ Alkanol \ \textbf{3} \ with \ Chiral \ 1,3-Disubstituted \ Allenes \ \textbf{2}^a)$

^a) Aldehyde **1** (1.3 mmol) and (i-Pr)₂Zn (2.7 mmol) were added in four portions. Molar ratio: allene **2**/ pyrimidine-5-carbaldehyde **1**/(i-Pr)₂Zn 0.0094/1.0/2.0. ^b) The ee value was determined by HPLC analysis with a chiral stationary phase. ^c) Molar ratio: allene **2a**/pyrimidine-5-carbaldehyde **1**/(i-Pr)₂Zn 0.037/1.0/2.0. ^d) Toluene was used instead of methylcyclohexane. Aldehyde **1** (1.1 mmol) and (i-Pr)₂Zn (2.2 mmol) were added in three portions. Molar ratio: allene **2a**/pyrimidine-5-carbaldehyde **1**/(i-Pr)₂Zn 0.024/1.0/2.0.

Conclusions. – As described, chiral allenes induce asymmetry in the enantioselective addition of $(i-Pr)_2Zn$ to 2-(alkynyl)pyrimidine-5-carbaldehyde **1**. Highly enantiomerically enriched pyrimidin-5-yl alkanol **3** was obtained by combination asymmetric autocatalysis. In addition, chiral allenes 2a - e are hydrocarbon compounds without any heteroatom. Thus, these results are also significant in that chiral hydrocarbon compounds [8e,g] without any stereogenic center act as chiral initiators in asymmetric synthesis.

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Experimental Part

Synthesis of Enantiomerically Enriched Allenes. Racemic allenes were synthesized according to the procedure in [11]. The racemic allenes obtained were resolved into enantiomerically enriched form by HPLC with a chiral stationary phase.

(+)-(S)- and (-)-(R)-1,3-Diphenylpropa-1,2-diene (2a) [3a]. Racemic 2a was synthesized [11] and resolved into enantiomerically enriched form by HPLC (*Chiralcel OD* (4×250 mm); eluent: 1% i-PrOH in hexane; flow rate: 0.4 ml/min): t_R 12 min for (-)-isomer, 18 min for (+)-isomer.

(+)-1-Cyclohexyl-3-phenylpropa-1,2-diene (**2b**) [3g]. $[a]_D^{28} = +299$ (c = 0.47, EtOH) for the sample with 90% ee. HPLC (*Chiralcel OD-H* (4×250 mm); eluent: 0.01% i-PrOH in hexane; flow rate: 0.4 ml/min): t_R 15 min for (–)-isomer, 18 min for (+)-isomer.

(-)-1-Cyclohexyl-3-phenylpropa-1,2-diene (**2b**) [3g]. $[a]_{D}^{29} = -345$ (c = 0.39, EtOH) for the sample with 92% ee.

(+)-(S)-4-Methyl-1-phenylpenta-1,2-diene (**2c**) [3a]. $[a]_{D}^{32} = +139$ (c = 0.33, CHCl₃) for the sample with 68% ee ([3a]: $[a]_{D}^{20} = +345$ (EtOH)). HPLC (*Chiralcel OD*-H (4 × 500 mm), eluent: 0.01% i-PrOH in hexane; flow rate: 0.4 ml/min): t_{R} 29 min for (–)-isomer, 32 min for (+)-isomer.

(-)-(R)-4-Methyl-1-phenylpenta-1,2-diene (2c) [3a]. $[a]_{D}^{32} = -277$ (c = 0.40, CHCl₃) for the sample with 98% ee.

(+)-1,4-Diphenylbuta-1,2-diene (2d). HPLC (Chiralcel OD (4 × 250 mm), eluent: 0.001% i-PrOH in hexane, flow rate: 0.5 ml/min, r.t.) $t_{\rm R}$ 27 min for (-)-isomer, 31 min for (+)-isomer. Colorless oil. $[a]_{\rm D}^{29}$ = +105 (c = 0.64, EtOH) for the sample with 38% ee. FT-IR (neat): 1948. ¹H-NMR (300 MHz, CDCl₃): 3.51 (dd, J = 7.3, 2.5, 2 H); 5.75 (td, J = 7.3, 6.5, 1 H); 6.20 (dt, J = 6.5, 2.5, 1 H); 7.2–7.4 (m, 10 H). ¹³C-NMR (75 MHz, CDCl₃): 36.0; 94.9; 95.4; 126.7; 127.1; 127.2; 128.9; 129.0; 135.0; 140.5; 206.1. HR-MS (FAB⁺): 206.1085 (M⁺; C₁₆H₁₄; calc. 206.1096.

(-)-1,4-Diphenylbuta-1,2-diene (2d). $[a]_{D}^{29} = -211$ (c = 0.23, EtOH) for the sample with 86% ee.

(+)-1,5-Diphenylpenta-2,3-diene (2e). HPLC (Chiralcel OD ($4 \times 250 \text{ mm}$), eluent: 0.5% 2-propanol in hexane, flow rate: 0.5 ml/min): t_R 14 min for (+)-isomer, 16 min for (-)-isomer. Colorless oil. $[a]_D^{25} = +1.9$ (c = 1.75, EtOH) for the sample with 97% ee. FT-IR (neat): 1963. ¹H-NMR (300 MHz, CDCl₃): 3.3–3.4 (m, 4 H); 5.2–5.3 (m, 2 H); 7.1–7.3 (m, 10 H). ¹³C-NMR (75 MHz, CDCl₃): 35.6; 91.0; 126.1; 128.3; 128.6; 140.2; 205.1. Anal. calc. for C₁₇H₁₆: C 92.68, H 7.32; found: C 92.47, H 7.06.

(-)-1,5-Diphenylpenta-2,3-diene (2e). $[a]_{D}^{25} = -4.87$ (c = 1.84, EtOH) for the sample with 92% ee.

Representative Procedure for the Enantioselective Addition of $(i-Pr)_2Zn$ to Pyrimidine-5-carbaldehyde in the Presence of Chiral Allenes (Table, Entry 2). To a methylcyclohexane (1 ml) soln. of (+)-(R)-1,3-diphenylpropa-1,2-diene (**2a**, > 99.5% ee; 9.6 mg, 0.05 mmol) and pyrimidine-5-carbaldehyde **1** (9.4 mg, 0.05 mmol) was added dropwise a 1M hexane soln. of $(i-Pr)_2Zn$ (0.15 ml, 0.15 mmol) over a period of 30 min at 0°. After stirring for 12 h at 0°, the mixture was diluted with toluene (1.9 ml). A toluene soln. (1M) of $(i-Pr)_2Zn$ (0.2 ml, 0.2 mmol) and a toluene (1.0 ml) soln. of **1** (18.8 mg, 0.1 mmol) were added successively at 0°. After stirring the mixture for 30 min at 0°, toluene (7.5 ml), a toluene soln. (1M) of $(i-Pr)_2Zn$ (0.8 ml, 0.8 mmol), and a toluene (2.0 ml) soln. of **1** (75.3 mg, 0.4 mmol) were added successively. Then, the mixture was stirred for further 30 min at 0°. Toluene (15 ml), toluene soln. (1M) of $(i-Pr)_2Zn$ (1.6 ml, 1.6 mmol), and a toluene (4.0 ml) soln. of **1** (150.6 mg, 0.8 mmol) were added successively, and the mixture was stirred at 0° for 30 min. The reaction was quenched by adding 5 ml of 1M HCl, and the resulting mixture was neutralized with sat. aq. soln. of NaHCO₃ (15 ml). The mixture was filtered through *Celite*. The filtrate was extracted with AcOEt and dried (Na₂SO₄). The solvent was removed, and the residue was purified on TLC (hexane/AcOEt 2:1): (S)-1-[2-[2-(tert-Butyl)ethenyl]pyrimidin-5-yl]-2-methylpropan-1-ol (**3**) with 98% ee was obtained in 95% yield (296.4 mg).

REFERENCES

- S. Patai, 'The Chemistry of Ketenes, Allenes, and Related Compounds', John Wiley & Sons, New York, 1980; S. R. Landor, 'The Chemistry of the Allenes', Academic Press, London, 1982; H. F. Schuster, G. M. Coppola, 'Allenes in Organic Synthesis', John Wiley & Sons, New York, 1984.
- [2] C. J. Elsevier, in 'Methods of Organochemistry', 4th ed., Ed. G. Helmchen, R. W. Hoffmann, J. Mulzer, E. Schaumann, Thieme Verlag, Stuttgart, 1995, Vol. E 21a, pp. 537–566.
- [3] a) C. J. Elsevier, P. Vermeer, J. Org. Chem. 1989, 54, 3726; b) A. Alexakis, I. Marek, P. Mangeney, J. F. Normant, J. Am. Chem. Soc. 1990, 112, 8042; c) T. Mukaiyama, M. Furuya, A. Ohtsubo, S. Kobayashi, Chem. Lett. 1991, 989; d) M. Aso, I. Ikeda, T. Kawabe, M. Shiro, K. Kanematsu, Tetrahedron Lett. 1992, 39,

5787; e) Y. Nishibayashi, J.-D. Singh, S.-i. Fukuzawa, S. Uemura, *J. Org. Chem.* **1995**, *60*, 4114; f) I. Ikeda, K. Honda, E. Osawa, M. Shiro, M. Aso, K. Kanematsu, *J. Org. Chem.* **1996**, *61*, 2031; g) A. G. Myers, B. Zheng, *J. Am. Chem. Soc.* **1996**, *118*, 4492; h) K. A. Reynolds, M. G. Finn, *J. Org. Chem.* **1997**, *62*, 2574; i) M. Franck-Neumann, D. Martina, D. Neff, *Tetrahedron: Asymmetry* **1998**, *9*, 697; j) H. Ohno, A. Toda, Y. Miwa, T. Taga, N. Fujii, T. Ibuka, *Tetrahedron Lett.* **1999**, *40*, 349; k) T. Satoh, Y. Kuramochi, Y. Inoue, *Tetrahedron Lett.* **1999**, *40*, 8815; l) Z. Wan, S. G. Nelson, *J. Am. Chem. Soc.* **2000**, *122*, 10470; m) D. J. Fox, J. A. Medlock, R. Vosser, S. Warren, *J. Chem. Soc., Perkin Trans. 1*, **2001**, 2240.

- [4] P. Stang, A. E. Learned, J. Org. Chem. 1989, 54, 1781; W. de Graaf, J. Boersma, G. van Koten, C. J. Elsevier, J. Organomet. Chem. 1989, 378, 115; K. Tanaka, K. Otsubo, K. Fuji, Tetrahedron Lett. 1996, 37, 3735; P. H. Dixneuf, T. Guyot, M. D. Ness, S. M. Roberts, Chem. Commun. 1997, 2083; Y. Naruse, H. Watanabe, Y. Ishiyama, T. Yoshida, J. Org. Chem. 1997, 62, 3862; K. Mikami, A. Yoshida, Angew. Chem., Int. Ed. 1997, 36, 858; Y. Noguchi, H. Takiyama, T. Katsuki, Synlett, 1998, 543; A. Tillack, C. Koy, D. Michalik, C. Fisher, J. Organomet. Chem. 2000, 603, 116; Z. K. Sweeney, J. L. Salsman, R. A. Andersen, R. G. Bergman, Angew. Chem., Int. Ed. 2000, 39, 2339; M. Oku, S. Arai, K. Katayama, T. Shioiri, Synlett 2000, 493; C. Schultz-Fademrecht, B. Wibbeling, R. Fröhlich, D. Hoppe, Org. Lett. 2001, 3, 1221; J. W. Han, N. Tokunaga, T. Hayashi, J. Am. Chem. Soc. 2001, 123, 12915.
- [5] O. W. Gooding, C. C. Beard, D. Y. Jacson, D. L. Wren, G. F. Cooper, J. Org. Chem. 1991, 56, 1083; E. M. Carreira, C. A. Hastings, M. S. Shepard, L. A. Yerkey, D. B. Millward, J. Am. Chem. Soc. 1994, 116, 6622; M. S. Shepard, E. M. Carreira, J. Am. Chem. Soc. 1997, 119, 2597; M. Node, K. Nishide, T. Fujiwara, S. Ichihashi, Chem. Commun. 1998, 2363; V. M. Arredondo, S. Tian, F. E. McDonald, T. J. Marks, J. Am. Chem. Soc. 1999, 121, 3633; J. D. Ha, J. K. Cha, J. Am. Chem. Soc. 1999, 121, 10012; J.-F. Poisson, J. F. Normant, J. Am. Chem. Soc. 2001, 123, 4639; J. A. Marshall, M. M. Yanik, J. Org. Chem. 2001, 66, 1373; R. K. Dieter, H. Yu, Org. Lett. 2001, 3, 3855.
- [6] K. Soai, T. Shibata, H. Morioka, K. Choji, *Nature* 1995, 378, 767; T. Shibata, H. Morioka, T. Hayase, K. Choji, K. Soai, *J. Am. Chem. Soc.* 1996, 118, 471; T. Shibata, S. Yonekubo, K. Soai, *Angew. Chem., Int. Ed.* 1999, 38, 659; I. Sato, D. Omiya, K. Tsukiyama, Y. Ogi, K. Soai, *Tetrahedron: Asymmetry* 2001, 12, 1965; I. Sato, T. Yanagi, K. Soai, *Chirality* 2002, 14, 166.
- [7] a) K. Soai, *Enantiomer* 1999, 4, 591; b) K. Soai, T. Shibata, I. Sato, *Acc. Chem. Res.* 2000, *33*, 382; c) K. Soai, I. Sato, T. Shibata, *Chem. Rec.* 2001, *1*, 321; d) C. Bolm, F. Bienewald, A. Seger, *Angew. Chem., Int. Ed.* 1996, *35*, 1657; e) M. Avalos, R. Babiano, P. Cintas, J. L. Jiménez, J. C. Palacios, *Chem. Commun.* 2000, 887; f) H. Buschmann, R. Thede, D. Heller, *Angew. Chem., Int. Ed.* 2000, *39*, 4033; g) C. Girard, H. B. Kagan, *Angew. Chem., Int. Ed.* 1998, *37*, 2922; h) B. L. Feringa, R. A. van Delden, *Angew. Chem., Int. Ed.* 1999, *38*, 3418; i) K. Mikami, M. Terada, T. Korenaga, Y. Matsumoto, M. Ueki, R. Angeland, *Angew. Chem., Int. Ed.* 2000, *39*, 3532.
- [8] a) T. Shibata, J. Yamamoto, N. Matsumoto, S. Yonekubo, S. Osanai, K. Soai, J. Am. Chem. Soc. 1998, 120, 12157; b) K. Soai, S. Osanai, K. Kadowaki, S. Yonekubo, T. Shibata, I. Sato, J. Am. Chem. Soc. 1999, 121, 11235; c) I. Sato, K. Kadowaki, K. Soai, Angew. Chem., Int. Ed. 2000, 39, 1510; d) S. Tanji, A. Ohno, I. Sato, K. Soai, Org. Lett. 2001, 3, 287; e) I. Sato, R. Yamashima, K. Kadowaki, J. Yamamoto, T. Shibata, K. Soai, Angew. Chem., Int. Ed. 2000, K. Soai, H. Ogino, Chem. Commun. 2001, 1022; g) I. Sato, S. Osanai, K. Kadowaki, T. Sugiyama, T. Shibata, K. Soai, Chem. Lett. 2002, 168.
- [9] G. Lowe, Chem. Commun. 1965, 411; P. Crabbé, E. Velarde, H. W. Andersons, S. D. Clark, W. R. Moore, A. F. Drake, S. F. Mason, Chem. Commun. 1971, 1261; R. Rossi, P. Diversi, Synthesis 1973, 25; C. J. Elsevier, P. Vermeer, A. Gedanken, W. Runge, J. Am. Chem. Soc. 1985, 107, 2537; W. Smadja, Chem. Rev. 1983, 83, 263.
- [10] J. M. Walbrick, J. W. Wilson, W. M. Jones, J. Am. Chem. Soc. 1968, 90, 2895; S. F. Mason, G. W. Vane, Tetrahedron Lett. 1965, 1593.
- [11] W. von E. Doering, K. Hoffmann. J. Am. Chem. Soc. 1954, 76, 6162; W. von E. Doering, P. M. LaFlame, Tetrahedron, 1958, 2, 75.

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